

## Novel Carbene Complexes of Rhodium

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**Summary** Stable diphenylcarbene complexes of rhodium (II, III, IV, and V) have been prepared.

VARIOUS transition metal carbene complexes<sup>1-4</sup> in which the carbene ligand is stabilized by an adjacent oxygen or nitrogen atom, have been reported. We now report the preparation of the first diphenylcarbene-rhodium com-

plex (II) is air stable and insoluble in most organic solvents except for bridge-splitting reagents such as pyridine in which it is converted, on recrystallization from  $\text{CH}_2\text{Cl}_2$ -petroleum ether, into red needles of (III)  $[\text{Rh}(\text{CO})(\text{CPh}_2)\text{Cl}(\text{C}_5\text{H}_5\text{N})_2] \cdot (\text{CH}_2\text{Cl}_2)_2$ . Complex (III) gradually decomposes over  $100^\circ$ ,  $\nu_{\text{C}=\text{O}}$   $1836\text{ cm}^{-1}$ ,  $\nu_{\text{Rh}-\text{Cl}}$   $296\text{ cm}^{-1}$ , mol.wt.  $(\text{CH}_2\text{Cl}_2)$  822 (calc. 822). Reaction of (II) with cyclopentadienylsodium in THF, after chromatography on an alumina column, gave the air stable, green crystals (IV),  $(\pi\text{-C}_5\text{H}_5)_2\text{-Rh}_2(\text{CPh}_2)_2(\text{CO})$ , 75% yield, m.p.  $179\text{--}182^\circ$  (decomp.),  $\nu_{\text{C}=\text{O}}$   $1840\text{ cm}^{-1}$ ,  $M^+$  696.

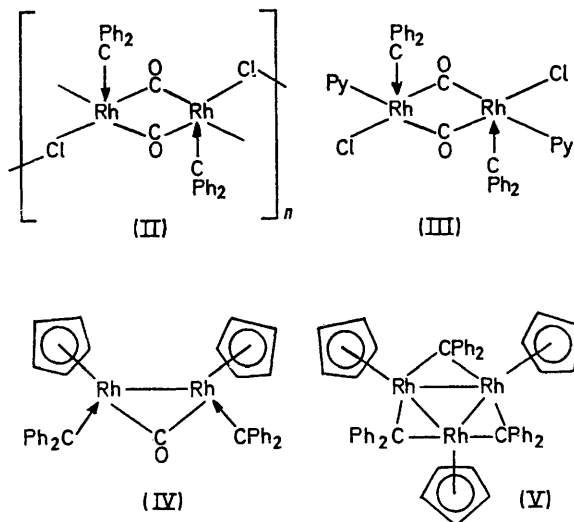
Complex (IV) slowly lost a bridging carbonyl on heating or if kept at room temperature in benzene, the trinuclear complex (V),  $(\pi\text{-C}_5\text{H}_5)_3\text{Rh}_3(\text{CPh}_2)_3$ , violet crystals, m.p.  $227\text{--}228^\circ$ ,  $M^+$  1002 was obtained.

Complex (II) when treated with triphenylphosphine under reflux in toluene gave  $\text{Ph}_2\text{C}=\text{CPh}_2$  (VI),  $\text{Rh}(\text{CO})(\text{PPh}_3)_2\text{Cl}$  (VII), and  $[\text{Rh}(\text{PPh}_3)_2\text{Cl}]_2$  in yields of 77%, 42%, and 33%, respectively. Although our attempt to trap diphenylcarbene in the above reaction using diphenylethylene at  $150\text{--}160^\circ$  has been unsuccessful resulting in the formation of (VI) and (VII) in the yields of 56% and 87%, respectively, the formation of (VI) supports the presence of a diphenylcarbene ligand. A similar carbene intermediate has been suggested<sup>5</sup> in the catalytic decarbonylation of diphenylketen with  $\text{Co}_2(\text{CO})_8$ .

On the basis of the above reaction and the i.r. and  $^1\text{H}$  n.m.r. spectra of (II), (III), (IV), and (V), we suggest the structures shown.

The structures of (IV) and (V) are related to those of  $(\pi\text{-C}_5\text{H}_5)_2\text{Rh}_2(\text{CO})_3$ <sup>6</sup> and  $(\pi\text{-C}_5\text{H}_5)_3\text{Rh}_3(\text{CO})_3$ ,<sup>7</sup> respectively. Further studies on other carbene complexes and X-ray studies of the complexes (III), (IV), and (V) are in progress.

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plexes.  $[\text{Rh}(\text{CO})_2\text{Cl}]_2$  (I) (6 mmol), was treated with diphenylketen (18 mmol) in xylene for 4 h under reflux and gave a 75% yield of an insoluble brick red complex (II),  $[\text{Rh}(\text{CO})(\text{CPh}_2)\text{Cl}]_n$ , m.p.  $290^\circ$  (decomp.),  $\nu_{\text{C}=\text{O}}$   $1860\text{ cm}^{-1}$ ,  $\nu_{\text{Rh}-\text{Cl}}$   $275\text{ cm}^{-1}$ . Complex (II) and its diarylcarbene analogues have also been obtained from (I) and diaryldiazomethane in benzene in high yields.

† Satisfactory analytical data for all formulated compounds have been obtained.

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